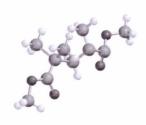
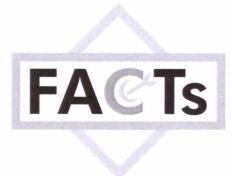
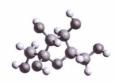
FRONTIER AREAS IN CHEMICAL TECHNOLOGIES – 2019

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EDITORS

- G. PARUTHIMAL KALAIGNAN
- T. STALIN

• G. GOPU

• S. VISWANATHAN





ALAGAPPA UNIVERSITY

(A State University Established in 1985) Karaikudi - 630003, Tamil Nadu, India











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Frontier Areas in Chemical Technologies – 2019 (FACTs - 2019)

² Department of Marine and Coastal Studies, Madurai Kamaraj University, Madurai, Tamilnadu, India.

E-mail: archanadev@gmail.com

Abstract

Lead (Pb) is one of the most widely distributed metals in the marine ecosystem. The acute toxicity tests of *D. faba* exposed to lead under acute Continuous Flow Through (CFT) method was conducted. After 24 h exposure, 100% survival of test animal was in all five lead exposed concentrations and after 48 h exposure, mortality was observed in the concentrations - 1.7, 3.4, 6.8 and 13.6mg.l⁻¹ and as compared to control animal, sudden reduction in protein level was noticed in the lowest concentration (0.85 mg.l⁻¹) 52% and in the highest concentration (13.6 mg.l⁻¹) the reduction was 69%. More damage of gill and foot tissue was observed in Pb exposed D. faba. In the highest concentration of 13.6 mg. 1⁻¹, coagulated gill filaments with disintegrated secretary vesicles found along the margin and also within the gill membrane. In the lowest concentration (0.85 mg l⁻¹), the activation of enzyme was only 17%. however the activation was 172% when D. faba was exposed in the highest concentration (13.6 mg l⁻¹). This present study indicates that *D. faba* is a marine bivalve molluscs species deserving of greater attention in future. Meanwhile, this could be regarded as a suitable bio indicator of Cd, Cu, Pb and Mn exposure by measuring Protein and LPO as biomarkers. LPO was considered to be an important feature in cellular injury. The reactive oxygen forms (O₂) H₂O₂,OH) which are produced during oxidative stresses potentially damaging cells and tissues.

Keywords: Lead, Accumulation, Tissue, Lipid peroxidation, Histology.

PP 10

A NOVEL SCHIFF BASE LIGAND AND ITS COMPLEXES FOR SENSOR APPLICATION

V. Aruldeepa^a, P. Tharmaraj^{b*} and C. D. Sheela^c

^{a*}Department of Chemistry, Fatima college, Madurai

^bPG and Research Department of Chemistry, Thiagarajar College, Madurai

^cPG and Research Department of chemistry, The American College, Madurai

Abstract

The versatile ligational behaviour of Schiff base compounds had evoked considerable interest in the past. It is our aim to synthesize new azo methine derivative and expected to Department of Industrial Chemistry, Alagappa University, Karaikudi.

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embit variety of characteristics such as biological and catalytic properties. We have monstrated that metal ions acting as modulators (or inputs, in digital design parlance) can merate absorbance changes in accordance with the operation of a half-adder. It was also mained exploiting differential binding affinities of metal ions for different ligands. The sent work focuses on the synthesis of ligand HPBABF and its complexes of copper, zinc, and nickel. All the synthesized compounds were characterized using various physical chemical methods of analysis such as UV-Vis, FTIR, NMR, fluorescence, CV, TG, SEM, and elemental analysis.

words: Hydroxypyrimidine, antimicrobial, metal complexes.

PP 11

SYNTHESIS AND ELECTROCHEMICAL CHARACTERIZATION OF HIGH VOLTAGE CATHODE MATERIAL (LiCo_{0.5}Mn_{1.5}O₄) FOR LITHIUM ION **BATTERIES**

N. Azhagumeenal^a. N. Radha^a & S. T. Nishanthi^b

PG & Research Dept. of chemistry, Alagappa Govt. Arts College, Karaikudi, Tamilnadu, India. ^bCSIR-CECRI, Karaikudi, Tamilnadu, India.

E-mail: chemradha74@gmail.com

Abstract

Cathode material LiCo_{0.5}Mn_{1.5} O₄ of was successfully synthesized by as sol-gel method. The cathode material was calcinated different temperature 120 0 C, 600 0 C, 900 0 The XRD result reveals Spinel compound of LiCo_{0.5}Mn_{1.5}O₄ for 600 0 C-12 hrs LiCo_{0.5}Mn_{1.5}O₄ for 600 0 C-12 hrs is well crystallized and lattice parameter, space group, cell volume is evaluated. The surface morphology, particle size and elemental analysis were analysis the SEM and EDAX spectrum. SEM result shows the smooth morphology and less particle size and EDAX shows weight percentage elements is better for LiCo_{0.5}Mn_{1.5}O₄ at 900 0 C -5hrs compared than LiCo_{0.5}Mn_{1.5}O₄ at 600 0 C -12hrs. The oxidation state is analysis by XPS and the highest binding energy occur that LiCo_{0.5}Mn_{1.5}O₄ at 900 0 C -5hrs compare than LiCo_{0.5}Mn_{1.5}O₄ at 600 0 C - 12hrs. Half cells were assembled and tested at C/10 rate. The maximum discharge capacity of around 106 mAh/g was obtained for the sample LiCo_{0.5}Mn_{1.5}O₄ at 900 0 C -5hrs. The charge transfer resistance value of LiCo_{0.5}Mn_{1.5}O₄ at 900 0 C -5hrs is 411.341 Ω also better compare than 600 0 C at 12 hrs

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