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Affiliated to Madurai Kamaraj University

Re-Accredited with 'A++' by NAAC (Cycle - IV)

Mary Land, Madurai - 625018, Tamil Nadu

PROGRAMME OUTCOMES AND COURSE OUTCOMES

2022 - 2023

NAME OF THE PROGRAMME: M.Sc Chemistry

PROGRAMME CODE: PSCH

Programme Outcomes:

PO1	Firm hold and sound footing in theoretical and practical aspects of Chemistry
PO2	An overall comprehensive and an in-depth knowledge and equip learners to possess global competency
PO3	Diversified branches with deep rooting cultivate research aptitude that leads to innovative findings
PO4	Informative but application oriented inputs
PO5	Enhanced chances to take up careers in industries and other pivotal sector.
P06	Rigorous training to tackle challenges in the academic and societal need based fields
PO7	Opportunity to be exposed to the current emerging trends in the field of Chemistry through activities such as workshops, seminars and projects.



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Course Outcomes:

Course Code	Course Title	Course Outcomes
19PG1C1	Inorganic	CO1: To analyse all chemical species involved in organic and Inorganic reactions and to identify those as acid and bases.
	Chemistry-I	CO2: To classify the bonds as ionic and covalent and to compare the theories.
		CO3: To categorize the solid systems, to calculate the lattice energy and draw conclusions on their stability.
		CO4: To predict the structures and magnetic properties of Inorganic compounds.
		CO5: To gainindepth knowledge of nuclear reactions, reactors and the applications of radio isotopes in all fields.



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19PG1C2	Organic Chemistry-I	CO1: To interpret the concept of aromaticity and the main properties of aromatic compounds.
		CO2: To explore reactivity patterns of conjugated ,aromatic molecules and to evaluate the kinetics and thermo dynamics controlled reactions.
		CO3: To define the fundamentals of chirality, prochirality, symmetry elements and applications of atropisomers.
		CO4: To comprehend of nucleophiles,
		electrophiles, electronegativity, and
		resonance
		CO5: To sketch the preparation and properties of heterocyclic compounds.
19PG1C3	Physical Chemistry-I	CO1: To gain knowledge Kohlrausch's law and electrolytic conductance, Calculate the molar conductance ,degree of dissociation and electrical potential Possess thorough understanding of Debye-Huckel equation.



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		CO2: To gain knowledge of Electro catalysis and Electro synthesis.
		CO3: Describe in detail about the three laws of thermodynamics.
		CO4: Restate in their own words about the concept of distribution, thermodynamic probability and most probable distribution CO5: Correlate and explain the partial molar properties, chemical potential
19PG1C4	Inorganic Qualitative Analysis	CO1: To study the principle of distribution of common and rare metal ions in different groups.
		CO2: To know the inter- and intra group precipitation and separation of metal ions.
		CO3: To improve the skill in the qualitative analysis of rare metal ions in different groups.
		CO4: To identify the methodology to analyse a metal ion in the presence of another metal ion.



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19PG1C5	Organic Qualitative	CO1: To be skilled in the separation of binary organic mixtures.
	Analysis& Preparation-I	CO2: To gain knowledge on the skills of doing micro level analysis.
		CO3: To know the methods of qualitative analysis of organic compounds
		CO4:To learn about the preparation of suitable derivative of the organic functional groups
		CO5: To prepare organic compounds.
21C1EDC	Analysis Of Soil, Water, Food,Cosmetic s And Oil	CO1: Acquire the complete knowledge of soil and its texture CO2: Develop idea about water and its treatment
		CO3: Identify different types of food colour, additives and food adulterants
		CO4: Learn the ingredients required for the preparation of various types of



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		shampoos, skin powder, nail polish.
		CO5: Understand the need of detoxification of oil and various adulterants present in oil.
19PG2C6	Inorganic Chemistry-II	CO1: Compare the stabilities of complexes using stability constants and to identify the types of isomers
		CO2: To describe the theories of co-ordination compounds to understand the colours and magnetic properties and their position in the spectrochemical series
		CO3: Investigate the structures of complexes using IR,NMR, ESR and other spectral techniques.
		CO4: Possess a thorough understanding of electronic spectra of complexes.
		CO5: To arrive at the mechanisms of



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		substitution reactions in six and four
		coordinated complexes using kinetic studies
		CO1: To comprehend the mechanism of elimination and substitution
19PG2C7	Organic Chemistry-II	reactions and to apply the stereochemistry in E1, E2,ionic and pyrolytic
	Chemistry ii	eliminations.
		CO2: To interpret the concept of nucleophilic and free radical addition
		reactions and metal hydride reduction and to discriminate the reactivity of
		organometalicre agents.
		CO3: To explore reactivity patterns of Substituted cyclohexanes and to employ conformational reactivity in cis and transdecalins and to apply conformations in SN1, SN2, ionic, pyrolytic eliminations and NGP reactions.
		CO4: To acquire a complete knowledge of the principles of UV, IR
		spectroscopy and toexamine the various functional groupspresent in
		organic molecules using λmax and IR frequency values .



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		CO5: To differentiate the molecular rearrangements and to solve the simple problems and to recall the various naming reactions and to interpret the products.
19PG2C8	Physical Chemistry-II	CO1: To acquire knowledge about the basic concepts of chemical kinetics CO2: To identify and analyze the effect of physical parameters μ , D on rate of reaction CO3: To derive rate constant for reactions using Lindemann, Hinshelwood, RRK, RRKM Theories. CO4: To develop a knowledge and understanding of the concept Normalisation and orthogonalisation and to solve Schrodinger wave equation for particle in a one dimensional box, three dimensional box and Rigid rotator. CO5: To apply variation and perturbation method to He atom.



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131 62 63	Inorganic Quantitative Analysis	CO1: To enable the students to acquire the quantitative skills in volumetric analysis and gravimetric analysis
		CO2: To improve the skill in quantitative estimation of metal ions by various titric methods.
		CO3: To identify the methodology to estimate a metal ion in the presence of another metal ion.
		CO4: To be skilled in synthesis of inorganic complexes.
19PG2C10	Organic Estimation & Preparation-II	CO1: To develop the ability for synthesizing organic compounds by single stage. CO2: To develop the ability for synthesizing Organic compounds by double stage. CO3: To study the reaction mechanism.
		CO1: Acquire the complete knowledge ofsoil and its texture.
21C2EDC	Analysis Of	



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	Soil, Water, Food, Cosmetic s And Oil	CO2:Develop idea about water and its treatment.
		CO3: Identify different types of food colour, additives and food adulterants.
		CO4: Learn the ingredients required for the preparation of various types of shampoos, skin powder, nail polish.
		CO5: Understand the need of detoxification of oil and various adulterants present in oil.
19PG3SICI	Internship	CO1: To carry out scientific experiments
		CO2: To accurately record and analyze the results of such experiments.
19PG3C11	Organic Chemisty- <i>III</i>	CO1: To acquire a complete knowledge of the basic principles of 1H-NMR,



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		13C-NMR and Mass spectroscopy
		CO2: To be acquainted with complete knowledge of photochemistry of ketone &cyclo addition reactions and to develop an understanding of the significance of the number, and splitting of signals in NMR
		CO3: To be competent to assign structures to simple molecules on the basis of nuclear magnetic resonance spectra
		CO4: To distinguish the similarities and differences of Pericyclic reactions and Cyclo addition and sigmatropic reactions
		CO5: To apply the Spectral concepts to solve the problems, to elucidate the structures of simple organic compounds using the data from all the spectral techniques
19PG3C12	Physical Chemistry-III	CO1: To learn about symmetry elements and symmetry operations, the point groups and character table



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		CO2: To Describe the selection rule for infrared-active and Raman active transitions, electronic transitions
		CO3: To analyse the hybridization of given compounds and to apply HMO theory to Ethylene and some conjugated systems
		CO4: To Classify of surface active agents, Polymers, and to derive Gibbs adsorption and BET isotherms
		CO5: To explain the kinetics of vinyl, cationic and anionic polymerizations and to determine the mass of polymers.
19PG3C13	Green Chemistry	CO1: To know about the alternative feedstock and to study about the process and advantages of alternative materials
		CO2: To get familiarise about the green chemistry technology



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		CO3: To understand the need of alternative energy sources
		CO4: To learn different types of renewable energy sources
		CO5: To acquire knowledge about the greener techniques in industries
19PG3CE1	Material Chemistry	CO1: To gain knowledge about the basic principles of nanochemistry and classification of nanomaterials.
		CO2: To describe several synthesis of inorganic nanoparticles, one-dimensional nanostructures (nanotubes, nanorods, nanowires), thin films, nanoporous materials, and nanostructured bulk materials,
		CO3: To criticize the importance of various instrumentation techniques such as NMR, IR, UV, X-ray diffraction, ESR etc., for elucidating the



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		structures of nanomaterials.
		CO4: To depict the structure of carnonnanoatructures, organic nanopolymers and supra molecular structures
		CO5: To recognize the important role of nanomaterials in various fields.
19PG3CE2	Bio-Organic Chemistry	CO1: Understand concepts of molecular recognition and drug design
		CO2: Remember the synthesis and structure of Proteins and amino acids.
		CO3: Know the extraction and purification of enzymes and their application in catalysis.
		CO4: Categorize and analyze enzyme mechanisms.
		CO5: Analyze the structure and biological functions of Coenzymes.



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19PG4C14	Physical Chemistry Practicals-I	CO1: Developed expertise relevant to the professional practice of chemistry
		CO2: Developed an understanding of the breadth and concepts of physical
		chemistry
		CO3: An appreciation of the role of physical chemistry in the chemical
		sciences and engineering
		CO4: Developed an understanding of the role of the chemist and chemical
		engineer in tasks employing physical chemistry
		CO5: An understanding of methods employed for problem solving in
		physical chemistry
19PG4C15	Inorganic Chemistry-III	CO1: Illustrate the structure and mode of bonding in organometallic complexes



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		CO2:Apply the different electron counting procedures to predict the shape and stability of organometallic complexes
		CO3:Illustrate the mechanism of dioxygen binding in various oxygen carrier proteins
		CO4: Classify and identify the different types of metallo enzymes and metallo proteins based on their biological functions.
		CO5:Interpret the structure of borazines, boranes and carboranes.
19PG4C16	Organic Chemistry-IV	CO1: To differentiate the carbon –carbon bond forming reactions and to interpret the products and to explore reactivity patterns of various coupling reactions
		CO2:To elucidate the structural units of quinine, morphine, □-pinene and □-codinene



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		CO3:To correlate the skeletal units of nucleotides and nucleosides- RNA and DNA
		CO4:To categorize the reducing and oxidizing agents and its applications.
		CO5:To Sketch the effective and logical synthetic route for the synthesis of new molecules
19PG4C17	Physical Chemistry-IV	CO1: Describe the structure and mode of bonding in organometallic complexes containing carbonyls, nitrosyls, carbenes, carbynes, alkenes, alkynes and also metallocene complexes
		CO2:Apply different electron counting procedures to predict the shape and stability of organometallic complexes
		CO3: Illustrate the mechanism of dioxygen binding in various oxygen carrier proteins



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		CO4: Classify different types of metallo enzymes and metallo proteins based on their biological functions.
		CO5:Distinguish whether the given compound belongs to chain or ring or cage or cluster
19PG4CE3	Analytical Chemistry	CO1: To acquire the complete knowledge of C language
		CO2: To develop logics which will help them to create programs, applications of chemistry problems in C.
		CO3:To explicate the theoretical principles of selected instrumental methods within electro analytical and spectrometric/spectrophotometric methods, and main components in such analytical instruments.
		CO4:To explain the confidence level and confidence limit, the sources of random errors and effects of random errors on analytical results.



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		CO5:To illuminate the theoretical principles of various separation techniques in chromatography, and typical applications of chromatographic techniques
19PG4CE4	Chemical Engineering	CO1: To write C- Program using various features of C- language
		CO2: To categorize the various conditioning methods in water treatment
		CO3:To apply the principles involved in spectrophotometric analysis.
		CO4:To compare the mechanism between dry corrosion and wet corrosion
		CO5:To synthesize some industrially important polymers
19PG4C18	Physical Chemistry Practicals- II	CO1: Experience in some scientific methods employed in basic and applied physical chemistry



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		CO2: Developed skills in procedures and instrumental methods applied in analytical and practical tasks of physical chemistry
		CO3: Developed skills in the scientific method of planning, developing, conducting, reviewing and reporting experiments
		CO4: Developed some understanding of the professional and safety responsibilities residing in working with chemical systems.
19PG4CPR	Project	CO1: To carry out scientific experiments
		CO2: To accurately record and analyze the results of such experiments.