FATIMA COLLEGE (AUTONOMOUS)



Re-Accredited with "A++" Grade by NAAC (4th Cycle)
Maryland, Madurai- 625 018, Tamil Nadu, India

NAME OF THE DEPARTMENT : PHYSICS

NAME OF THE PROGRAMME : Ph.D

PROGRAMME CODE : DSPH

ACADEMIC YEAR : 22-23

	Minutes of the Board of Studies
	Meeting
	To be implemented from 2022-2023 onwards
	Venue: A1
	Convered on 23-03-2022 at 2pm
Agris	Members Present:
1.	Dr. A. Sheela Vimala Rani Head of the Dept
	Dr. Basherrudin Mahmud Ahmed University Asst. Prof. School of Physics Nominee Madurai Kamaraj University Madurai
	Madurai Kamaraj University Madurai Ballo
3.	Dr. K. Marimuthu Subject Asst. Prof Escpert
	Asst. Prof Esepert Department of Physics Grandhigram Rival Instituti — Deemed University Grandhigram
	Gandhigram Deemed University
1.	Dr. M. Umando.i Associate Professor > Head Expert
	Department of Physics Mother Teresa Women's University 23/03/92
	Attiwam patti, Kodaikanal

Industrialist Industrial Electronics Cosporation No. 1, V. Ramputash Industrial Estate Madurai Alumnae 6. Dr. R. Vishnu Briya Dept. of Physics The Madura College Som feigraf Madurai 7. Dr. Malathy Asst. Prof. Dean of Academic Affair
23/3/2022 Dept. of Zoology Fatima College L. Caroline Sugnihan 8 Dr. L. Caroline Sugiithan Associate Prof. 9. Mrs. R. Alphonsa Fernando Associate Professor R. A. Lend hhulhendry. 10. Dr. M.V. Leena Chandra Asst. Prof. 1. Asst. Prof. 2. ty . L. Am Tysk 12 Dr. Ancemnia Joseph Asst. Prof.

Magamin. 13. Dr. M. Ragam Asst. Prof. 14. Dr. G. Jenita Rani Asst. Prof. of a ling R. Jo Jour 15. Dr. R. Tothimani Asst. Prof. 16. Ms. I. Janet Shorly Asst. Prof. Of Of 17. Ms. J. R. Sofia
Asst. Prof. J. P. 89.6 R.Nj-18 Dr. R. Niranjana Devi Bost. Prof. AGENDA FOR BOARD OF STUDIES 1. Preparation of Action taken report 2. To carry out atleast 5-10% changes in a minimum - 20% or more in the courses offered (Approximately 8-10 courses minimum) 20% - same code (b) courses with revision more than 20%. - New code - to be prefixed with 22... 3. New courses to be introduced · Course code to be prefixed with 22...

4. New Value - Added Courses can	20
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4. New Value - Added Courses can introduced or the titles can be changed	re
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5. Each department to offer at least one Value - Added Courses per y	ear
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		Solid State Physics paper	

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	energy sources"	introduced
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		Mathematical Physics	arra Marin
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		is a misnomer in	
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32	DECTAL DE	Quantum Mechanics	MARKET
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		- A. Shankar	Larathde 1
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		(11) Quantum Mechanics:	
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		applications -	
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500		Industrialistic suggested	
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les	Power	currently existing	2 100 2 db 20010

		DO PHYSIES
	SUGGESTIONS IN THE PREVIOUS BOARD	ACTION TAKEN IN THE ACADEMIC YEAR 2021-22
10.50	for advanced learners entitled on "Instrumentation and experimental methods" by paper entitled on "Digital Signal Processing"	This paper is to be passed in this board

2. REVISION OF COURSES:

S-No.	Course	Units	7. of	Course Title
	Code	Units revised	revision	19
1.	19 PLACCY	Unit is _ Bernoull	15%	Allied Physics-P
(3)	19M3ACPI/	theorem, unit II	MUNICALINA	Physics-P
	19G3ACPI	entropy, unit iv	La Browling	0
2.	1992 ACC3/	chars lains -	Samuel Can	
	19M4ACP2/	unit IT -		Allied
	19 G4 ACP2	Frank-Hertzegt		Physics 17
3	19P5CC14	unit i? - Lasers	15%.	Physics-17 OPTICS
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		Spectroscopy	enzlaan.	\
	Philips /	1 1/0/	andrea	
4	19P6cc17	Unit I -	5/.	Thermodynami
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		included		
5	19850016	Non-electro	44	Revised
		Praeticals	1 Coas in	los.

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Ŀ				10%	Medical
4	L TEX		10 17 0 //	Resear	Paysics
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1	7		unit & = Seniconductor	101	Applied
			memories included		Electronics
	8		Unit 7 - Quantum	Sauga	Nuclear
5	<u>breë</u>		Electrodynamics,	10%	and Partic
No.	5,00.0	10	SU3 symmetry		Physics
			included		0

3 NEW COURSES INTRODUCED:

	PROGRAM	COURSE	COURSE
ſ.	B.Sc.	22P4CC11	Mathematical Methe
2.	B.Sc.	22P4SB2	Solar Cell and its
3	M.Se.	22PRSL2	Balteries and its
4	M.Sc.	22 PGSLPI	Applications Digital Signal Processing
25%	Carter to A	MANAGER FUN	rucessing

4. NEW VALUE ADDED COURSE:

Course Code	Course Tible
22 PGVAPCI	PG Diploma in
	Instrumentation on Electrochemical Workstation

5. Approval of Ph.D. Course Work Syllabrus: Course Work paper and Core paper for the Research Scholar are as follows:

Ph.D. Scholar	Course work paper	Core paper
P. Mohanaa	22PHDCWPOI	22PHDCPP02
Mutauselii	Solid State Jonics	Materials Science
	* L . L . A . T	

6. SUGGESTIONS GIVEN BY THE BOARD MEMBERS:

the new course Mathematical Methods can be offered as elective/skill based paper

* The nomenclature for the above paper can be changed ento Numerical Methods

* A paper on Mathematical Physics can be included as core paper which will form a cusis to standying project.

Solid State Physics etc.

* "Interference" can be shifted from
Allied Physics-II to Allied Physics-I
to be on par with the practicals

* The Board suggested to include "Principles of Electronics" by Mehta as one of the reference looks. * In the electrice paper (19P6ME1), instead of Timer and Counter Assemblers and Compilers can be included. The Applied - Fixed man payed PG * The nomenclature for the New Value added course was discussed in detail. * The above course can be offered as "certificate course". * So the little of the New Value added course is certificate course on Instrumentation Electrochemical techniques. * In the Nuclear and Particle Physics course, the Board

suggested to include "Nuclear Physics - Theory and Experiments" by Roy End Nigam as reference book. * In the Applied Electronics paper, the Industrialist suggested that Pulse width Modulation and Suitching regulators can be introduced instead of registers and counters. * Mr. Ramprakash also recommended to wichide Assemblers and Simulators in the course "Instrumentation and Microcontroller" 21PG2P10 and to reduce the content of programming in 8051. * The subject experts strongly recommended to specify the details of sections in the books for Istudy in all the units of the syllatrus for both us and PG programs.

The specifications of the sections would enable the students to learn more precisely. The detailed sections in all units would fecilitate the examiner also:

* CREDIT TRANSFER OF SWAYAM MOOC COURSE:

The board members suggested the credit transfer of Swayam -

- 1) Both the syllabris should be same
- 2) Number of hours should make

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1) Dr. A. Sheela Vimale	
2) Dr. Basherrudin Mahmud	Ahmod A. Bantos.
	Malante
4) Dr. M. Umaderi	Bruedey 103/02/22
5) Mr. Rampralcash	V-Rimpontoch
6) Dr. R. Vishner Priya	18mm 23/03/22
7) Dr. Malashy	Abalelli 23/3/22
8) Dr. L. Caroline Sugrikham	L'anoline Engritage
9) Mrs. R. Alphonsa Fernando	R. A. Lendo
10) Dr. M. V. Leener Chandra	Mhura Chem Dry
11) Mrs. I . Jeyasheela	2. Ty 1
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4) Dr. Sr. G. Jenita Rani	Fra Reig
15) Dr. R. Jothimani	R. Joyan
16) Ms. I. Janet Sherly	Blog.

17) Ms. J. R. Sofia 18) Dr. R. Niranjana Devi

COLLEGE PROFILE

Fatima College (Autonomous), Mary Land, Madurai, is a Post Graduate and Research Institution for Women affiliated to Madurai Kamaraj University. It is a Catholic Minority institution established and run by St. Joseph's Society of Madurai (of the Congregation of the Sisters of St. Joseph of Lyons, France). This institution came into existence through the tireless efforts of the missionary sisters of St. Joseph of Lyons and the zeal and heroic sacrifice of Rev. Sr. Rose Benedicta, the Foundress of the College.

The College was started in St. Joseph's Campus Madurai as a Second Grade College with 63 students in 1953. It was upgraded into a Post Graduate College in 1964; Autonomous in 1990 and a Research Institute in 2004. The College now offers 21 Undergraduate Programmes, 13 Postgraduate Programmes, 2 Professional Programme, 5 M.Phil. **Programmes** 6 Departments have become Research Centres. It has strength of 4134 Students, 206 Teaching Staff and 100 Non-Teaching Staff.

The comprehensive assessment by NAAC in 1999 placed Fatima College in Five Star Status of merit. The college strives to sustain excellence, quality and relevance while equipping the students to meet the demands of higher education in India. In 2004 UGC conferred on Fatima College the status of College with Potential for Excellence. In 2006 and 2013 NAAC Re-Accredited the College with 'A' Grade. The College was ranked 94th in the All India NIRF Ranking in 2019 by MHRD.

VISION

WOMEN'S EMPOWERMENT THROUGH EDUCATION

The vision of the college is to empower women by developing human capabilities through quality education based on Christian values, making them responsible citizens who can work for the advancement of the society and promote communal harmony in the multi-religious and multi-cultural reality of India eventually evolving into women of communion.

MISSION

- To enhance quality of life through the development of individuals.
- To enable women to become contributors in the economic, social and political development of India.
- To equip the students with 21st century skill-sets with a focus on problem-solving abilities
- To motivate them to work for social justice
- To give preference to the rural economically backward and first-generation learners
- To enable students to be employed in the technology oriented competitive market

VISION OF THE DEPARTMENT

Educate, Empower and Excel

MISSION OF THE DEPARTMENT

- To ignite the young minds and impart quality education in basic Physics
- To promote enthusiasm in the study of physics through innovative and dedicated teaching methodologies
- To discover the budding talents in theoretical and experimental physics and ensure their global competency
- To provide a stimulating environment and strengthen basic and application oriented research aptitude among the students.

FULL TIME DOCTOR OF PHILOSOPHY

<u>PHYSICS - Year - I</u> For those who joined in 2022 onwards

PROGR AMME CODE	COURSE CODE	COURSE	CATE GORY	HRS/WE EK	CREDITS
DSPH	22PHDCWP01	SOLID STATE IONICS	Core	-	2

COURSE DESCRIPTION

This course emphasises the basic concepts of Solid State Ionics which involves its structure, properties and its applications in electrochemical devices.

COURSE OBJECTIVES

This course provides detailed information about the field of Ionics and its contribution in electrochemical devices.

UNITS

Unit I: Ionic Conductors

Types of Ionic solids- Fast Ionics Solids-Point Defect type-Sub Lattice type – Fast Ionic materials – alkali metal-ion conductors - β aluminas- Silver ion conductors- Cation conductors- Oxygen ion conductors – Halide ion conductors – Proton conductors – Electronic conductors with ionic transport.

Unit II: Preparation Methods

Various methods of preparation of amorphous/glassy, poly and single crystalline materials – thermal evaporation– sputtering – glow – discharge decomposition –chemical vapour deposition – melt quenching – gel dissociation –crystal growth technique – X-ray diffraction and differential thermal analysis-Glass transition – factors determining glass transition temperatures – structure – microscopic structure – modelling– microscopic structure – examples.

Unit III: Transport Mechanism

Point Defect Type: Point defect type super ionic conductors – transport mechanism through defects – jumpfrequency – ionic conductivity and diffusion co-efficient – defect concentration – pure and doped crystals –impurity vacancy association – coulomb interactions-Application of transport theory to fluoride and Oxygen ion conductors: Molten Sub-lattice type: Molten Sub-lattice type solid state ionic conductors – Hypermann's theory –Rices Strassler & Toouch's theory – Welch Dieme's theory – Lattice gas theory – Path Probability and Moute Carlo Methods – Ionic Percolation theory – Jahn Teller Model-Dynamics- ion transport – free ion model – domain model – jump diffusion model and frequency dependent conductivity.

Unit IV: Characterization Techniques

Macroscopic properties – electrical conductivity – diffusion thermo electric power-Microscopic properties – X- ray diffraction studies – a.c. conductivity – dielectric relaxation – NMR – ESR – far IR – Mossbauer Spectroscopy –Raman Scattering – Photo Electron Spectroscopy – Cyclic Voltametry – cycle stability of electrodes and electrolytes.

Unit V: Electro Chemical Devices

Thermodynamic studies – general aspects of solid state batteries – electrolyte – compatibility between electrode substance and solid electrolytes – electrode structure –interface between electrode and solid electrolyte – Fuel cell-Proton exchange membrane fuel cell-

High temperature fuel cells – solid state potentiometer gauges for gaseous species – coulometer – electro-chemical capacitor- electro chromic display system.

Text Book:

- Anthony R. West, Solid State Chemistry and its Applications , Wiley, 2014
- 2. S. Chandra, Superionic solids: Principles and applications, Elsevier North-Holland, 1981
- 3. L. L. Hench and J. K. West, Principles of Electronic Ceramics, John Wiley & Sons, New York, 1990

BOOKS FOR REFERENCES:

- **1.** Solid state Ionics. (Eds. T Kudo and Fueki) VCH Publishers, Kodansha 1990.
- **2.** Lectures on solid state physics (Eds. G Bush and H Schade), international series on Natural Philosophy Vol. 79 Pergamon, press 1976.
- **3.** "Solid Electrolyte" (Eds. S Geller) Springer Verlag New york 1977.

COURSE OUTCOMES

On the successful completion of the course, students will be able to:

NO.	COURSE OUTCOMES	KNOWLEDGE LEVEL (ACCORDING TO REVISED BLOOM'S TAXONOMY)	PSOs ADDRESSED	
CO 1	understand the basic ideas on ion conductors	K1, K2	PSO1,PSO2	

CO 2	Explore the different preparation methods of ionic conductors	K1, K2, K3	PSO3,PSO4
CO 3	understand the transport mechanism of super ionic conductors	K1, K2	PSO1,PSO3
CO 4	discussion on various characterization techniques that can be used for the analysis of super ionic solids	K2, K2, K3& K4	PSO4, PSO5
CO 5	Study that involves the applications of solid state Ionics in various electrochemical devices	K1, K2, K3 & k4	PSO3,PSO4& PSO5

COURSE DESIGNER: Dr. M. V. Leena Chandra

Forwarded By

Dr. A. Sheela Vimala Rani HoD'S Signature & Name

FULL TIME DOCTOR OF PHILOSOPHY

PHYSICS - Year - I

PROG RAM ME CODE	COURSE CODE	COURSE TITLE	CATEG ORY	HRS/ WEEK	CREDI TS
DSPH	22PHDCPP02	MATERIALS SCIENCE	Ph.D Core	-	2

COURSE DESCRIPTION

This course aims at providing the theoretical aspects of materials science.

COURSE OBJECTIVES

This course provides the knowledge about phase diagrams, mechanical properties, ceramics, polymers, plastics and crystals.

UNITS

Unit -I: PHASE DIAGRAMS

Solid solutions and intermediate phases – Equilibrium phase diagrams, Cu-Ni, Pb-Sn, Al-Cu system phase diagrams – Free energy and equilibrium phase diagrams – Nucleation and growth – Martenstic transformation – Strengthening mehanisms – Iron Carbon system – Alloy steels – Aluminium-Copper system – Copper-Zinc system – Corrosion

Unit -II: MECHANICAL PROPERTIES

Stress- Strain curve – Elastic deformation: Characteristics, Atomic mechanism, Sheer stress, Bulk modulus, Strain energy, Strain

deformation – Viscous deformation: Spring-Dashpot models – Anelastic and Viscoelastic deformation: Viscoelastic models – Plastic deformation: Dislocations and Stress-strain curves, Plasticity theory – Fracture: Ideal fracture, Brittle fracture, Fracture mechanics, Cohesive models, Ductile fracture – Mechanical testing

Unit -III: CERAMICS

Structure of ceramics – Production of ceramics: Raw materials, Forming and Post-forming processes – Production of glass: Melting of glass, Glass forming and annealing – Mechanical properties of ceramics – Wear and erosion resistance – Thermal shock – Silica-Alumina system – Commercial systems: Zirconia, Sialones, Cement and Concrete

Unit - IV: Polymers and Plastics

Molecular structure: Monomers & Polymers, Synthesis, Molecular weight measurement, Branching & Tacticity, Copolymers and blend – Mechanics of polymer chain: Freely jointed chains, Entanglements, Rubber elasticity – Thermoplastic melts: Viscosity, Shear thinning, Processing, Extrusion – Amorphous polymers: Solidification, glass transition, Various models – Crystalline polymers – Crosslinked polymers: Elastromers, Thermosets – Liquid crystal polymers – Mechanical properties: Stress-Strain behavior – Chemical properties

Unit -V: Biomaterials

Classification of Amino acids - Zwitter ion formation and isoelectric point - Synthesis ofglycine, alanine, and phenyl alanine - Peptide bond. Synthesis of peptides - Classification of proteins - Primary, secondary and tertiary structure of proteins - Denaturation of proteins - Tests for proteins - Carbohydrates - Properties of glucose, fructose and sucrose - Cyclic structures and Haworth projections of glucose, fructose, maltose and sucrose - Mutarotation. - Structure of starch and cellulose. Nucleic acids: Structure of pentose sugar, nitrogenous base, nucleoside and nucleotide - Double-helical structure of DNA - Differences between DNA and RNA- Energy rich molecules: Elementary structure of ATP, ADP and AMP.

BOOKS FOR STUDY AND REFERENCES:

- J.C.Anderson, K.D.Leaver, P. Leevers and R.D.Rowlings, Materials Science for Engineers, Nelson Thomas Ltd, First Indian reprint, 2010.
- **2.** M.Arumugam, Materials Science, Anuradha Agencies, Publishers, Sechond Edition, Fifth Reprint, 2005.
- **3.** R,Balasubramaniam, Materials Science and Engineering, Wiley India (P) Ltd, 2010.
- **4.** V.Raghavan, Materials Science for Engineering, Prentice Hall of India Pvt Ltd, 2006.

COURSE OUTCOMES

On the successful completion of the course, students will be able to

NO.	COURSE OUTCOMES	KNOWLEDGE LEVEL (ACCORDING TO REVISED BLOOM'S TAXONOMY)	
CO 1	gain knowledge about the phase diagram	K1,K2	PSO1,PSO2
CO 2	know about the basic ideas of mechanical properties of the materials	K1, K2	PSO1,PSO2
CO 3	understand the basic concepts of ceramics	K1, K2, K3	PSO2, PSO3
CO 4	know about the polymers and plastics	K1, K2, K3& K4	PSO4,PSO5
CO 5	gain information about the basic knowledge of structure and chemical composition of biomaterials.	K1, K2 , K3 & K4	PSO4,PSO5

COURSE DESIGNER: Dr. M. V. Leena Chandra

Forwarded By

Dr. A. Sheela Vimala Rani HoD'S Signature & Name