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Affiliated to Madurai Kamaraj University

Re-Accredited with 'A++' (CGPA 3.61) by NAAC (Cycle - IV)

Mary Land, Madurai - 625018, Tamil Nadu

#### PROGRAMME OUTCOMES AND COURSE OUTCOMES

2021 - 2022

PROGRAMME CODE: UACH

NAME OF THE PROGRAMME: B.Sc Chemistry

**Programme Outcomes (POs)** 

PO 1	Are skilled in problem solving, critical thinking and analytical reasoning.
PO2	Are able to identify and solve chemical problems and explore new areas of research.
РО3	Are able to communicate the results of their work to chemists and non-chemists.
PO4	Students will be able to explain that chemistry is an integral part in addressing social, economic, and environmental problems.
PO5	Students turn out to be globally competent there by establishing themselves as attractive professionals.



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### **Course Outcomes (COs)**

Course Code	Course Title	Course Outcomes
19C1CC1	Inorganic Chemistry -I	CO1: To comprehend the fundamental properties of atoms, molecules, and the various states of matter
		CO2. To classify the electronic structure of atoms and its influence on chemical
		CO3. To acquire the knowledge of properties, characteristics and application of non-aqueous solvents
		CO4. To recognize the anomalous properties of Li and compares the properties Li with those other alkali metal
		CO5. To illustrate the factors affecting the strength of acid and bases.
19C1CC2	Organic Chemistry –I	CO1.Gain a thorough knowledge about the chemistry of aliphatic saturated compounds
		CO2. Analyze the behaviour of an organic compound through

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	(Reaction mechanism, alkanes, cycloalkanes and alkyl halides)	electron displacement effects  CO3. Describe the structure and stability of different types of intermediates involved in reaction mechanism.  CO4. Know the nomenclature, classification of alkanes, alkyl halides.  CO5. To derive and familiarise the mechanisms of nucleophilic substitution reactions of organic compounds.
19C1CC3	Volumetric Analysis-I	CO1. To compare the principles behind all types of titrations CO2. To identify suitable indicators for a particular reaction. CO3. To prepare solutions of desired concentrations.  CO4. To apply the principles of volumetric analysis in acid base, permanganometry, and iodometric titrations.



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19N1ACC1	Allied Chemistry-I	CO1. To predict the geometry of any molecule with the help of VB and VSEPR theory
		CO2. To construct M.O diagram for homonuclear diatomic molecule
		CO3. To categorize the types of organic reactions
		CO4. To describe the chemistry of carbohydrates.
		CO5. To classify the chemical reactions involved in volumetric analysis
19Z1ACC1	Allied Chemistry-I	CO1. To predict the geometry of any molecule with the help of VB and VSEPR theory
		CO2. To construct M.O diagram for homonuclear diatomic molecule
		CO3. To categorize the types of organic reactions
		CO4. To describe the chemistry of carbohydrates.



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		CO5. To classify the chemical reactions involved in volumetric analysis
19C1NME	Profitable	
	Home	CO1. Recognize the important nutrients present in food
	Industries	CO2. Gain knowledge about the fundamental chemistry involved in dairy products
		CO3. Determine the manufacture and functions of various soaps and creams
		CO4. Learn the ingredients required for the preparation of various types of shampoos, skin powder, nail polish
		CO5. Demonstrate the preparation of some home products like candle, detergent powder, soap oil,ink, phenoyl and computer sambirani



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19Z1ACC2	Allied Chemistry Practicals -I	CO1. Describe the principles and procedures of various titrimetric methods  CO2. Identify suitable indicators for a particular reaction  CO3. Know the various terms such as standard solution, normality, molality, molarity, equivalent weight and molecular weight.  CO4. Select the specific titrimetric method to estimate the amount of analyte present in the given solution  CO5. Apply the expressions and equations to calculate the strength of solutions
19N1ACC2	Allied Chemistry Practicals -I	CO1. Describe the principles and procedures of various titrimetric methods  CO2. Identify suitable indicators for a particular reaction  CO3. Know the various terms such as standard solution, normality, molality, molarity, equivalent weight and molecular



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		weight.  CO4. Select the specific titric method to estimate the amount of analyte present in the given solution  CO5. Apply the expressions and equations to calculate the strength of solutions
19C2CC4	Inorganic Chemistry –II (Theories of hard and soft acids – bases, chemical bonding and chemistry of group III, IV, V & VI Elements)	CO1. To categorize the soft, hard and border line acids and bases.  CO2. To compare Valence bond theory and molecular orbital theory  CO3. To understand the synthetic importance of organo metallic compounds of Al, B and Si  CO4. To criticize the chemistry of hydrazine and hydroxyl amine CO5. To draw the structure of oxo halides and oxo acids of sulphur.



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19C2CC5	Organic Chemistry –II (Alkenes, Alkynes, Alkadienes, organo metallic compounds, Alcohols and Ethers)	CO1. Gain a basic knowledge about elimination reactions to prepare alkenes  CO2. Describe the chemical reactions and structure of alkenes  CO3. Classify the alkadienes and alkynes  CO4. Choose the specific reagents to prepare various organic compoundsfrom GR  CO5. Compare the properties of alcohols and ethers
19C2CC6	Volumetric Analysis-I1	CO1. To apply the principles of volumetric analysis in various estimations.  CO2. To estimate the amount of calcium using permanganometric method  CO3. To estimate the amount of calcium and magnesium using



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Mary Land, Madurai - 625018, Tamil Nadu

		EDTA method.  CO4. To apply the principle of Argentimetry in the estimation of chloride ions.  CO5. To understand the principles behind the estimations of phenol & Aniline iodometrically.
19Z2ACC3	Allied Chemistry-II (Theory behind chemical bonding, and organic qualitative analysis, kinetics of chemical reactions and catalysis)	CO1. Apply the rules for naming the coordination complexes and to illustrate the applications of metal complexes in biological systems.  CO2. To analyze the various organic compounds qualitatively  CO3. To understand the procedure involved in detection of elements.  CO4. To explain the kinetics of a chemical reaction and to calculate the order of a particular reaction  CO5. To evaluate the types of catalysis and theories of catalysis

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19N2ACC3	Allied Chemistry-II (Theory behind chemical bonding, and organic qualitative analysis, kinetics of chemical reactions and catalysis)	CO1. Apply the rules for naming the coordination complexes and to illustrate the applications of metal complexes in biological systems.  CO2. To analyze the various organic compounds qualitatively  CO3. To understand the procedure involved in detection of elements.  CO4. To explain the kinetics of a chemical reaction and to calculate the order of a particular reaction  CO5. To evaluate the types of catalysis and theories of catalysis
19Z2ACC4	Allied Chemistry Practicals-II	CO1. Gain the knowledge of appearance, colour, physical state and odour of organic substances.  CO2. Distinguish whether the given compound is Aliphatic or



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		Aromatic, and Saturated or Unsaturated.
		CO3. Perform the confirmatory test for various functional groups present in the given organic compound.
		CO4. Recognize the usage of apparatus and laboratory reagents.
		CO5. Relate the experimental observations with theory behind practicals.
	Allied Chemistry	CO1. Gain the knowledge of appearance, colour, physical state
19N2ACC4	Practicals-II	and odour of organic substances.
		CO2. Distinguish whether the given compound is Aliphatic or Aromatic and Saturated or Unsaturated.
		CO3. Perform the confirmatory test for various functional groups present in the given organic compound.
		CO4. Recognize the usage of apparatus and laboratory



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		reagents.  CO5. Relate the experimental observations with theory behind practicals.
19C2NME	Profitable	CO1. Recognize the important nutrients present in food
	Home	CO2. Gain knowledge about the fundamental chemistry
	Industries	involved in dairy products
		CO3. Determine the manufacture and functions of various soaps and creams
		CO4. Learn the ingredients required for the preparation of various types of shampoos, skin powder, nail polish
		CO5. Demonstrate the preparation of some home products like candle, detergent powder, soap oil, ink, phenoyl and computer sambirani
19C3CC7	Organic	CO1. To interpret the concept of aromaticity and the main



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	and Inorganic Chemistry	properties of aromatic compounds.  CO2. To explore reactivity patterns of conjugated, aromatic molecules and to evaluate the kinetics and thermodynamics controlled reactions.  CO3. Explain types of oxides and oxyacids, their structure and reactivity in halogens  CO4. Discuss the properties d block elements & triads of
		transition elements.  CO5. Recognize the role of oxidizing agents, reducing agents, group reagents and complexing agents, and inferences with theory behind practicals.
19C3CC8	Physical Chemistry-I (Gaseous state,	CO1. Gain a basic knowledge about the kinetic theory of gases, gaseous laws, types of velocities and properties of gases CO2. Distinguish between ideal and non-ideal solutions CO3. Derive the relationship between molar mass of a non-



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Mary Land, Madurai - 625018, Tamil Nadu

19C3SB1	Solutions, dilute solutions, radio activity & Nuclear transformations and nuclear chemistry)  Agricultural Chemistry	volatile solute and colligative properties  CO4. Calculate the mass defect, packing fraction and binding energy for any nuclei  CO5. Predict the growing rate, mechanism and age of plants using radioactive elements  CO1. Define the term soil  CO2. Describe the various types of fertilizers and their uses  CO3. Realise the requirements of manures and fertilizers for better production of various types of crops  CO4. Examine the adverse effect of pesticides  CO5. Calculate the amount of calcium and magnesium present in various types of soils
19C3SB1(A)	Diary Chemistry	CO1.To understand The Composition, physical and chemical properties of milk.

Annual Quality Assurance Report (AQAR) (2021-2022)



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		CO2.To Know the minerals and vitamins present in the milk. CO3.To Gain the skills to develop milk powder processing CO4. To Gain knowledge about the chemistry of milk and milk products
19P3ACC1	Allied Chemistry-I (Theory behind chemical bonding, quantitative and qualitative analysis, kinetics of chemical reactions and thermodynamics)	CO1. To comprehend the fundamental theories of Valence Bond, types of overlapping and VSEPR.  CO2. To categorize the reactions involved in volumetric analysis CO3. To analyze the various organic compounds qualitatively CO4. To recognize the theories of chemical kinetics.  CO5. To highlight the importance of thermodynamics and its related functions.
19C3CC9	Inorganic Qualitative	CO1. Gain the knowledge of appearance, colour, physical state, and odour of inorganic substances



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	Analysis	CO2. Distinguish whether the given compound is interfering or non-interfering radicals.  CO3. Perform the confirmatory test for various acid and basic radicals present in the given inorganic compound.  CO4. Recognize the usage of apparatus and laboratory reagents.  CO5. Avoiding hazardous experiments by doing microlevel ecofriendly experiments.
19P3ACC2	Allied Chemistry Practicals-I	CO1. Describe the principles and procedures of various titrimetric methods
		CO2. Identify suitable indicators for a particular reaction
		CO3. Know the various terms such as standard solution, normality, molality, molarity, equivalent weight and molecular weight.
		CO4. Select the specific titric method to estimate the amount



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		of analyte present in the given solution.  CO5. Apply the expressions and equations to calculate the strength of solutions.
19C4CC10	Inorganic Chemistry-III	CO1.Know the structure and bonding of important coordination compounds
	(Coordination Chemistry)	CO2. Apply the rules to calculate the magnetic properties of complexes and how magnetic moments can be employed for the interpretation of their structure
		CO3.Get an overview about the reaction mechanism of metal complexes
		CO4.Import the skills to elucidate the structure and mode of bonding in organometallic compounds
		CO5.Gain knowledge about the chemistry of Lanthanides and Actinides



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Mary Land, Madurai - 625018, Tamil Nadu

19C4CC11	Physical Chemistry-II (Chemical Kinetics, Solid State and distribution Law)	CO1. To determine integrated rate expression for zero order, first order, second order reactions and their respective half-life period expressions with examples  CO2. To study the various factors which affect the rate of a chemical reaction such as concentration, temperature, and solvent  CO3. To learn the crystal diffraction and experimental techniques used to characterize the solid crystals  CO4. To recognize and give the lattice parameter relationships for the seven crystal systems  CO5. To value the Nernst distribution law - its thermodynamic derivation, modification of law when solute undergoes
		association, dissociation and chemical combination with one of the solvents
19C4SB2	Natural and Synthetic Dyes	CO1. Know and comprehend the principle and theories of dyes CO2. Identify the chromophoric groups and auxochromes present in the dyes

Annual Quality Assurance Report (AQAR) (2021- 2022)



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		CO3. Classify the of dyes whether natural or synthetic CO4. Predict the structure of dyes CO5. Recognise the applications of dyes in various industries
19C4SB2(A)	Health Chemistry	CO1.To Acquire the basic knowledge about the significances of food and hygiene
		CO2.To Classify the given drugs whether they belong to antipyretics, analgesics, depressants etc
		CO3. To Interpret the structure and mechanism of enzyme action
		CO4.To Catagorize and identify the function of the different types of harmones
		CO5. To Analyse the reason for common diseases affecting the human body



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19C4CC12	Organic Qualitative Analysis	CO1. Gain the knowledge of appearance, colour, physical state, and odour of organic substances  CO2. Distinguish whether the given compound is Aliphatic or Aromatic and Saturated or Unsaturated.  CO3. Perform the confirmatory test for various functional groups present in the given organic compound.  CO4. Recognize the usage of apparatus and laboratory reagents.  CO5. Avoiding hazardous experiments by doing microlevel eco friendly experiments.
19P4ACC3	Allied Chemistry-II (Periodic table and atomic properties, electro chemistry–I, II, Catalysis and	CO1. Understand the periodicity in periodic table CO2. Understand the different types of condutances and their relations and the effect of dilution. CO3. Use Nernst equation to calculate the electrode potential and emf of electrochemical cells. Study the applications of



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	photochemistry)	electrochemical measurements
		CO4. Understand the basics of photochemistry using laws of photochemistry and Jablonsky diagram
		CO5. Derive the rate constants o certain photochemical reactions.
19P4ACC4	Allied Chemistry Practicals	CO1. Gain the knowledge of appearance, colour, physical state and odour of organic substances.
		CO2. Distinguish whether the given compound is Aliphatic or Aromatic and Saturated or Unsaturated.
		CO3. Perform the confirmatory test for various functional groups present in the given organic compound.
		CO4. Recognize the usage of apparatus and laboratory reagents.
		CO5. Relate the experimental observations with theory behind



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19C5CC13	Organic chemistry –III (Aldehydes And Ketones, Carboxylic Acids And Their Derivatives, steroisomerism, Amines And Diazo Compounds And Carbohydrates)	CO1.To analyze the synthetic importance of reactive methylene compounds CO2.To generalize the characteristic features of optical isomers and geometrical isomers
	PHYSICAL CHEMISTRY –III	CO1. To predict the feasibility of chemical reactions applying II law of thermodynamics
19C5CC14	(Thermodynamics, Phase Rule &	CO2. To explain the absolute entropy of substances and to

Annual Quality Assurance Report (AQAR) (2021-2022)



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	Group Theory)	calculate it
19C5CC15	INORGANIC PRACTICALS (Gravimetric Analysis)	CO1. Acquire the knowledge of concept of gravimetric estimations.  CO2. Recognise the role of reagents in chemistry.
19C5CC16	GREEN CHEMISTRY PRACTICALS	CO1. Recognize the usage of apparatus and laboratory reagents.  CO2. Relate the experimental observations with theory behind practicals.
19C5ME1	SPECTROSCOPY	CO1. To identify various functional groups present in organic molecules using IR frequency.  CO2.To predict the number and nature of protons/ carbons in organic molecules in 1H-NMR/ 13C-NMR spectroscopy
19C5ME2	BIO CHEMISTRY	CO1.To identify the various metabolic reactions



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		CO2. To understand the importance of nucleic acids
19C5SB3	MEDICINAL CHEMISTRY	CO1.To study the mechanism of drug action CO2.To determine the designing and binding of drugs with receptors
19C5SB4	NANO CHEMISTRY	CO1. Learn about the background on Nanoscience.  CO2. Understand the synthesis of nanomaterials and their application and the impact of nanomaterials on environment
19C6CC17	Organic chemistry –IV  (Polynuclear Hydrocarbons, Heterocyclic Compounds, Amino Acids And Proteins)	CO1. To explicate the structures of Citral, Dipentene and Camphor.  CO2. To distinguish the properties of quinolin and isoquinoline.
19C6CC18	PHYSICAL	CO1. Calculate the cell potential for a nonstandard cell.

Annual Quality Assurance Report (AQAR) (2021-2022)



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	CHEMISTRY-IV  (Electrolytic Conductance And Electrochemistry)	CO2. Know the chemical reactions used in a lead-acid battery
19C6ME3	Advanced Organic Chemistry	CO1. To sketch Frontier molecular orbitals in photochemistry.  CO2. To differentitate the molecular rearrangements and to solve the simple problems
19C6ME4	Polymer Chemistry	CO1. To understand the theories and mechanism of different types of polymerisation processes.  CO2. To study the applications of the above techniques to synthesize different natural and synthetic polymers.
19C6ME5	Advanced Physical Chemistry	CO1. To understand the theories behind the spectral techniques like MW.IR,NMR and ESR CO2. To study the applications of the above techniques to elucidate the structures of molecules



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19C6ME6	Advanced Inorganic Chemistry	CO1.To understand the theories behind inorganic photochemistry and electroanalytical techniques.  CO2. To study the applications of the above techniques to elucidate the structures of Bio-inorganic molecules
19C6SB5	Computers In Chemistry	CO1. To write programs to determine lattice energy, half-life, normality, molarity, molality CO2. To present structure based drug designing in both 2D and 3D
19C6SB6	Green Chemistry	CO1. To differentiate between yield and atom economy CO2. To interpret the concept of Stereo selectivity, Chemo selectivity and Regio selectivity
19C6CC19	Physical Practicals	CO1. Experience in some scientific methods employed in basic and applied physical chemistry CO2. Developed skills in procedures and instrumental methods applied in analytical and practical tasks of physical chemistry