# FATIMA COLLEGE (AUTONOMOUS)



Re-Accredited with "A" Grade by NAAC (3<sup>rd</sup> Cycle) 74<sup>th</sup> Rank in India Ranking 2020 (NIRF) by MHRD Maryland, Madurai- 625 018, Tamil Nadu, India

NAME OF THE DEPARTMENT: CHEMISTRY

NAME OF THE PROGRAMME: M.Sc.,

PROGRAMME CODE : PSCH

ACADEMIC YEAR : 2021-2022

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5. Mr. S. Manikandan, Industrialist.
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## FATIMA COLLEGE (AUTONOMOUS)

Affiliated to Madurai Kamaraj University
Re-Accredited with 'A++' (CGPA 3.61) by NAAC (Cycle - IV)
Mary Land, Madurai - 625018, Tamil Nadu

#### **DEPARTMENT OF CHEMISTRY**

For those who joined in June 2019 onwards (For the academic year 2021-2022)

PROGRAMME CODE: PSCH

COURSE CODE	COURSE TITLE	HR S / W K	CREDI T	CIA Mk s	ES E Mk s	TOT MKs
	SEMESTER	– I				
19PG1C1	INORGANIC CHEMISTRY-I (Basic concepts, covalent and ionic bonding, solid state and crystallography, and Nuclear chemistry)	6	4	40	60	100
19PG1C2	ORGANIC CHEMISTRY-I (Reaction mechanism and stereochemistry)	6	4	40	60	100
19PG1C3	PHYSICAL CHEMISTRY-I (Applied electro chemistry & Statistical thermodynamics)	6	4	40	60	100
19PG1C4	INORGANIC QUALITATIVE ANALYSIS	4	2	40	60	100
19PG1C5	ORGANIC QUALITATIVE ANALYSIS	4	2	40	60	100
21C1EDC	ANALYSIS OF SOIL, WATER, FOOD, COSMETICS AND OIL	3	3	40	60	100
	LIBRARY	1		-		
Total		30	19			

	SEMESTER -	- II				
19PG2C6	INORGANIC CHEMISTRY-	6	4	40	60	100
	(Advanced coordination chemistry)					
19PG2C7	ORGANIC CHEMISTRY-II	6	4	40	60	100
	(Elimination and addition reactions, organic spectroscopy and conformational analysis)					
19PG2C8	PHYSICAL CHEMISTRY- II	6	4	40	60	100
	(Chemical kinetics and Quantum mechanics)					
19PG2C9	INORGANIC QUANTITATIVE ANALYSIS	4	2	40	60	100
19PG2C10	ORGANIC QUANTITATIVE ANALYSIS	4	2	40	60	100
21C2EDC	ANALYSIS OF SOIL, WATER, FOOD, COSMETICS AND OIL	3	3	40	60	10 0
	LIBRARY	1		-	-	-
Total		30	19			
	SEMESTER -	III	T	1	1	
19PG3SIC1	INTERNSHIP/SUMMER PROJECT*	-	3	50	50	100
19PG3C11	ORGANIC CHEMISTRY- III	6	5	40	60	100
	(Spectroscopy and Pericyclic reactions)					
19PG3C12	PHYSICAL CHEMISTRY- III	6	5	40	60	100
	(Group Theory, Surface Chemistry and Macromolecules)					
19PG3C13	GREEN CHEMISTRY	6	5	40	60	100
19PG3CE1 /	MATERIAL CHEMISTRY / BIO ORGANIC	4	4	40	60	100

19PG3CE2	CHEMISTRY					
19PG3C14	PHYSICAL CHEMISTRY	6	4	40	60	100
	PRACTICALS-I					
	(Electrical Experiments-I)					
	LIBRARY	2				
Total		30	26			
	SEMESTER -	IV				
19PG4C15	INORGANIC CHEMISTRY-					
	III (Organometallics	6	5	40	60	100
	&					
	Bio-inorganic chemistry)					
19PG4C16	ORGANIC CHEMISTRY- IV (Retrosynthesis,		-	4.0	60	100
	Reactions and Reagents,	6	5	40	60	100
	Natural					
	Products)					
19PG4C17	PHYSICAL CHEMISTRY- IV	6	5	40	60	100
	(Spectroscopy, Kinetic					
	Theory of gases,					
	Photochemistry And Radiation chemistry)					
19PG4CE3	ARIAINTOAI	4	4	40	60	100
1	ANALYTICAL CHEMISTRY /	4	4	40	60	100
19PG4CE4	CHEMICAL					
10001515	ENGINEERING PHYSICAL CHEMISTRY					
19PG4C18	PRACTICALS-II	6	4	40	60	100
	(Non-electrical experiments)					
19PG4CPR	PROJECT*& VIVA VOCE	-	3	40	60	100
	LIBRARY	2				
Total		30	26			
	Total	120	90			

## **OFF-CLASS PROGRAMME**

## ADD-ON COURSES

Cours e Cod e	Courses	Hrs ·	Cred its	Semest e r in which the course is offered	CIA Mk s	ES E Mk s	Total Mark s
	SOFT SKILLS	40	4	I	40	60	100
	COMPUTER APPLICATIONS	40	4	II	40	60	100
	MOOC COURSES (Department Specific Courses) * Students can opt other than the listed course from UGC-SWAYAM /UGC /CEC	_	Minim u m 2 Credit s	-	-	1	
	COMPREHENSI VE VIVA (Question bank to be prepared for all the papers by the respective course teachers)	-	2	IV	-	-	100
	READING CULTURE	15 / Se me ste r	1	I-IV	-	-	-
	TOTAL		13 +				

## • EXTRA CREDIT COURSE

#### • Lab Courses:

o A range of 10-15 experiments per semester

#### • Summer Internship:

o Duration-1 month (2<sup>nd</sup> Week of May to 2<sup>nd</sup> week of June-before college reopens)

#### • Project:

- o Off class
- Evaluation components-Report writing + Viva Voce (Internal marks-50) + External marks 50

#### • EDC:

Syllabus should be offered for two different batches of students from other than the parent department inSem-I &Sem-II

# SELF LEARNING COURSE: OFFERED BY DEPARTMENT OF CHEMISTRY

COURSE	Course TITLE	H r s	Credi ts	Semes ter in which the course is offere d	CIA Mks	E S E M k s	Tot al Mar ks
21PG2SLC	RESEARCH METHODOLOGY	-	2	II	40	60	100

#### SEMESTER - II

#### For those who joined from 2021 onwards

PROGRAM	COURSE	COURSE TITLE	CATEG	HRS/WE	CREDI
ME CODE	CODE		ORY	EK	TS
PSCH	21C1EDC/ 21C2EDC	ANALYSIS OF SOIL, WATER, FOOD, COSMETICS AND OIL	EDC	3	3

#### **COURSE DESCRIPTION**

This paper focuses on all the important aspects of theory about soil, water, food chemistry, cosmetics and oil.

#### **COURSE OUTCOME:**

After completion of the course the students should be able to:

- Acquire the complete knowledge ofsoil and its texture
- Develop idea about water and its treatment
- Idetify different types of food colour, aditives and food adulterants
- Learn the ingredients required for the preparation of various types of shampoos, skin powder, nail polish.
- Understand the need of detoxification of oil and various adulterants present in oil.

#### **COURSE OBJECTIVES**

To understand the concepts of soil texture, water analysis.
To acquire the basic knowledge about food colour, food
additives and food and adulterants.
To learn sources of oil, analysis of oil and adulterant in oil.

#### UNITS

#### UNIT -I SOIL

(9 HRS.)

Composition of soil: Organic and Inorganic constituents. Soil acidity: buffering capacity of soils. Liming of soil. Absorption of cations and anions: availability of soil nutrients to plants.

#### UNIT-II WATER

(9 HRS.)

Importance of water.Naturalwater.Sources of water. Drinking water – making water fit to drink – chlorination. Water pollution-Chemicals causing water contamination – contamination by fertilizers, soaps and detergents and their effect.

#### UNIT-III FOOD CHEMISTRY

(9HRS.)

Food- composition of food -Color- Natural colouring matters – chlorophylls – carotenoids -Synthetic colours – permitted colours-banned colours - FPO, FSSAC, Agmark – Flavors - Food additives-Food adulterants and their detection in various food items.

#### **UNIT-IV COSMETICS**

(9HRS.)

Dental preparations-Tooth paste-Ingredients, their characteristic functios- Soap-hard soap and soft soap- Hair care preparations-Shampoo Shampoos – different kinds of shampoos –anti dandruff, anti-lice, herbal and baby shampoos -Hair dye –manufacture of conditioners -skin preparation –skin powder, nail polish, lipsticks.

#### UNIT -V OIL

(9 HRS.)

Natural sources of oils and fats, oils rich in palmitic acid and stearic acid- processing of fats and oils- analysis of oils- technical refining of oils for industrial uses- detoxification- shelf life prediction test-adulterants in oils.

#### **REFERENCES:**

- G.T. Austin : shreve's Chemical Process Industries, 5th edition,
   Mc- Graw-Hill, 1984
- Lakshmi, S. Pharmaceutical Chemistry, S. Chand and Sons, New Delhi, 1995.
- 3. A.K. De, Environmental Chemistry, New Age International Publishers, 2018.
- 4. JayashreeGhosh, Fundamental concepts of Applied chemistry, S.Chand publications, New Delhi (2013).
- 5. J.V.Simons, Science and Beauty Business Vol-1, Macmilan Education

Ltd, 1989.

#### **COURSE CONTENTS & LECTURE SCHEDULE:**

Module No.	Topic	No. of Lectures	Teaching Pedagogy	Teaching Aids
	UNIT -1 TITL	E - SOIL		
1.1	Composition of soil	1	Chalk & Talk	Black Board
1.2	Organic and Inorganic constituents	1	Chalk & Talk	Black Board
1.3	Soil acidity	1	Chalk & Talk	Black Board
1.4	buffering capacity of soils	1	Chalk & Talk	PPT & White board
1.5	Liming of soil	1	Chalk & Talk	Black Board
1.6	Absorption of cations	1	Chalk & Talk	Black Board
1.7	Absorption of anions:	2	Chalk & Talk	PPT & White board

	CBCS Curriculum for l	PG Chemis	stry				
1.8	availability of soil nutrients to plants.	SOIL HITTIANTS TO		Chalk & Talk	Black Board		
	UNIT - 2 TITLE -WATER						
2.1	Importance of water	1		Chalk & Talk	Black Board		
2.2	Naturalwater	1		Chalk & Talk	Black Board		
2.3	Sources of water	1		Chalk & Talk	Black Board		
2.4	Drinking water	1	C	halk & Talk	Black Board		
2.5	Making water fit to drink – chlorination	1	C	halk & Talk	PPT & White board		
2.6	Water pollution	1	С	halk & Talk	Black Board		
2.7	Chemicals causing water contamination –	1	С	halk & Talk	Black Board		
2.8	contamination by fertilizers, soaps and detergents and their effect	2	D	emonstration	Various raw materials		
UNIT -	3 TITLE -FOOD CHEMISTRY			,			
3.1	Food- composition of food	1		Chalk & Talk	Black Board		
3.2	Food colour	1		Chalk & Talk	Black Board		
3.3	Natural colouring matters chlorophylls – carotenoids	1		Chalk & Talk	Black Board		
3.4	Synthetic colours	1		Chalk & Talk	Black Board		
3.5	permitted colours	1	Chalk & Talk		Black Board		
3.6	banned colours - FPO, FSSAC, Agmark – Flavors	1		Chalk & Talk	Black Board		

3.7	Food additives	1	Chalk & Talk	Black Board
3.8	Food adulterants and their detection in various food items.	2	Chalk & Talk	Black Board

UNIT -4 TITLE-COSMETICS						
4.1	Dental preparations-Tooth paste- Ingredients, their characteristic functios		Chalk & Talk	Black Board		
4.2	Soap-hard soap and soft soap	1	Chalk & Talk	LCD		
4.3	Hair care preparations-Shampoo different kinds of shampoos –anti dandruff and anti-lice	2	Chalk & Talk	Black Board		
4.4	herbal and baby shampoos	1	Chalk & Talk	Black Board		
4.5	Hair dye –manufacture of conditioners	1	Chalk & Talk	Black Board		
4.6	skin preparation –skin powder	1	Chalk & Talk	Black Board		
4.7	nail polish	1	Chalk & Talk	Black Board		
4.8	lipsticks	1	Chalk & Talk	Black Board		
UNIT - 5	5 TITLE -OILS					
5.1	Natural sources of oils and fats	1	Chalk & Talk	Black Board		
5.2	oils rich in palmitic acid and static acid	1	Chalk & Talk	LCD		
5.3	processing of fats and oils	1	Chalk & Talk	Black Board		

5.4	analysis of oils	2	Chalk & Talk	Black Board
5.5	technical refining of oils for industrial uses	1	Chalk & Talk	Black Board
5.6	detoxification	1	Chalk & Talk	Black Board
5.7	shelf life prediction test	1	Chalk & Talk	Black Board
5.8	adulterants in oils	1	Chalk & Talk	Black Board

	C1	C2	C3	Total Scholastic Marks	Non Scholastic Marks C4	CIA Total	
Levels	Weekly	Monthly	MID- SEM TEST				% of Assessment
	5Mks.	10 Mks.	20 Mks.	35 Mks.	5 Mks.	40 Mks.	
K1	-	5 Mks.	5 Mks.	10	-	10	25 %
K2	-	5 Mks.	8 Mks.	13	-	13	32.5 %
К3	5 Mks.	-	7 Mks.	12	-	12	30 %
Non	-	-	-	-	5	5	12.5 %
Scholastic							
Total	5	10	20	35	5	40	100 %

CIA		
Scholastic	35	
Non Scholastic	5	
	40	

- ✓ All the course outcomes are to be assessed in the various CIA components.
- ✓ The levels of CIA Assessment based on Revised Bloom's Taxonomy for I PG are :

**K1**-Understand, **K2**-Apply, **K3**-Analyse

✓ The I PG course teachers are requested to start conducting S1, W1, M1,

## **EVALUATION PATTERN**

SCHOLASTIC			NON - SCHOLASTIC		MARKS		
C1	C2	С3	C4	C <sub>5</sub>	CIA	ESE	Total
5	10	15	5	5	40	60	100

**C1** – Average of Two Session Wise Tests

**C2** – Average of Two Monthly Tests

C3 - Mid Sem Test

C4 – Best of Two Weekly Tests

**C5** – Non - Scholastic

## **COURSE OUTCOMES**

On the successful completion of the course, students will be able to:

NO.	COURSE OUTCOMES	KNOWLEDGE LEVEL (ACCORDING TO REVISED BLOOM'S TAXONOMY)	PSOs ADDRESSED
CO 1	Analyse the buffering capacity of soil, p H, cation exchange capacity, nutrient availability of soil, fertility status of soil.	K1	PSO1
CO 2	Analyze the p H of water, hardness of water and acquire knowledge of advanced water purification techniques (and water treatment)		PSO2
СО 3	Identify different types of food colour,additives and food adulterants	K1	PSO2
CO 4	Learn the ingredients required for the preparation of the various types of shampoos,skin powder and nail polish	K2	PSO4
CO <sub>5</sub>	Analyze and Detect the presence of adulderants in oils and to compare the physical and chemical refining of oils	К3	PSO5

# **Mapping COs Consistency with PSOs**

CO/PSO	PSO 1	PSO 2	PSO 3	PSO 4	PSO 5	PSO 6	PSO 7	PSO 8	PSO 9
CO1	3	1	2	3	2	2	3	3	2
CO2	3	2	1	2	1	2	2	2	2
CO3	2	3	1	2	2	2	2	2	2
CO4	2	1	2	3	2	2	2	2	2
CO5	2	2	2	2	3	2	2	2	2

CO/ PSO	PO1	PO2	PO <sub>3</sub>	PO <sub>4</sub>
CO <sub>1</sub>	3	2	2	2
CO <sub>2</sub>	3	2	2	2
CO <sub>3</sub>	3	3	3	3
CO <sub>4</sub>	3	2	3	2
CO <sub>5</sub>	3	3	2	3

♦ WeaklyCorrelated -1

## **COURSE DESIGNER:**

- 1. Mrs. RM. Nagalakshmi
- 2. Dr. B.SUGANTHANA

**Forwarded By** 

**HOD'S Signature.** 

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## Curriculum for M.Sc. Chemistry

### I M.Sc., SEMESTER -1

## For those who joined in 2019 onwards

PROGRA MME CODE	COURSE CODE	COURSE TITLE	CATEGO RY	HRS/WE EK	CREDITS
PSCH	21PG2SLC	Research Methodology	PG Self learning	-	2

#### **COURSE DESCRIPTION**

This paper focuses on all the important aspects of Research Methodology

#### **COURSE OBJECTIVES**

This course helps the students to study about all concepts related to Research problem, literature survey, Web and library resources for research and writing research papers and proposals.

#### **Course Outcomes (COs)**

CO1	Introduce the purpose and importance of research.
CO2	Understand the various sources of information for literature
	survey.
CO3	Illustrate the Web and library resources for research.
CO4	Understand the writing of research papers &know the
	methodology of writing thesis and journal articles.
CO5	Analyse the writing of research proposal.

#### **UNIT - 1: Introduction to Research**

The search for knowledge, purpose of research, scientific method, characteristics of research, Types of research- fundamental or pure research, applied research, action research, historical research, experimental research.

Explanation of research problems, sources of research problems, selection of research problem characteristics of a good research problem, errors in selecting a research problem.

#### **UNIT-II: Literature Survey**

Sources of information, Primary, Secondary, Tertiary sources, Journals, Journal abbreviations, Abstracts, Current titles, Reviews, Monographs, Textbooks, Current contents, Introduction to Chemical Abstracts. Online searching, Database, *Scifinder, Scopus*, Citation Index, Impact Factor.

#### **UNIT-III:** Use of Web resources

The Internet and World Wide Web, internet resources for chemistry, internet search engines, using spreadsheets, word processors, databases and other packages, finding and citing information.

#### **UNIT-IV: Scientific Writing**

General aspects of scientific writing, reporting practical and project work, Format of the research report, style of writing the report, references and bibliography, Steps to publish a scientific article in a journal: types of publications- communications, articles, reviews; when to publish, where to publish, specific format required for submission, organization of the material, abbreviations used in scientific writing.

**UNIT-V: Writing of Research Proposal:** Research Proposal: Format of research proposal, individual research proposal and institutional proposal.

#### **Reference Books:**

- 1. Ranjit kumar, Research Methodology: A Step by Step Guide for Beginners, Pearson Education; 2<sup>nd</sup> Ed., (2005).
- 2. Dr.C.R. Kothari, Research Methodology: Methods and Techniques, New Age International Publishers, 2<sup>nd Ed.,</sup> New Delhi (2014.)
- 3. M.D. Barbara Gastel and Robert A. Day, How to Write and Publish a Scientific Paper, Greenwood Publishing Group Inc, 8th Ed., 2016.

- 4. Tanmoy Chakraborty and Lalita Ledwani, Research Methodology in Chemical Sciences: Experimental and Theoretical Approach, Apple Academic Press; 1st Ed.,, 2016.
- 5. R. L. Dominoswki, Research Methods, Prentice Hall, 1981.
- 6. H. F. Ebel, C. Bliefert and W. E. Russey, The Art of Scientific Writing, VCH, Weinheim, 1988.
- 7. H. M. Kanare, Writing the Laboratory Notebook; American Chemical Society: Washington, DC, 1985.
- 8. J. S. Dodd, Ed., The ACS Style Guide: A Manual for Authors and Editors; American Chemical Society: Washington, DC, 1985.
- 9. Gibaldi, J. Achtert, W. S. Handbook for writers of Research Papers; 2nd ed.; Wiley Eastern, 1987.
- 10. Joseph, A. Methodology for Research; Theological Publications: Bangalore, 1986

#### **PSO**

	Equip with an in-depth knowledge of varied fields namely			
PSO 1				
	Organic Chemistry, Inorganic Chemistry, Physical and			
	nanochemistry.			
PSO 2	Train in problem solving procedures enables to interpret the			
	experimental data into structures and mechanisms.			
PSO 3	Provides a tremendous exposure and cultivates analytical and			
	synthesising measures necessary to take up project work in			
	reputed institutions.			
PSO 4	Programme renders diversified thinking thereby promotes			
	creative skills.			
PSO 5	to solve the problems that cause a negative impact on			
	surroundings to pursue salient steps to safeguard environment			
PSO 6	Application-oriented input sharpens the skill to undertake CSIR-			
	NET exam.			
PSO 7	Knowledge with practical dimensions becomes a driving power to			
	undertake research in different areas at a global level.			
PSO 8	Multi-layered input enables to avail opportunities at chemical,			
	pharmaceutical industries.			
PSO 9				
	Becomes a contributing force and development agent in society.			
μ				

# Mapping of COs with PSOs

CO/ PSO	PSO 1	PSO 2	PSO 3	PSO 4	PSO 5	PSO 6	PSO 7	PSO 8	PSO 9
CO1	1	2	3	3	1	1	3	2	1
CO2	2	1	1	3	1	1	3	2	1
CO3	1	2	1	3	1	1	3	2	1
CO4	3	3	3	3	1	1	3	2	1
CO5	1	3	3	3	1	1	3	2	1

Mapping of COs with POs

CO/ PSO	PO1	PO2	РО3	PO4	PO5	P06	PO7	PO8	PO9
CO1									
CO2									
CO3									
CO4									
CO5									

**Note**: ♦ Strongly Correlated – **3** ♦ Moderately Correlated – **2** 

♦ Weakly Correlated -1

#### **COURSE DESIGNER:**

1. Dr.S.Sukumari

Forwarded By

**HOD'S Signature**